

B.Sc. DEGREE (C.B.C.S.S.) EXAMINATION, MARCH 2019**Sixth Semester****Core Course—CONDENSED MATTER PHYSICS**

(Common for B.Sc. Physics Model I, Physics Model II, Physics EEM and Physics Instrumentation)

[2013 Admission onwards]

Time : Three Hours

Maximum Marks : 60

Part A

Answer all questions.

Each question carries 1 mark.

1. The solid composed of atoms or other microscopic particles arranged in an orderly repetitive array is called _____.
2. The number of possible space lattices is _____.
3. The nature of bonding in sodium chloride is _____.
4. The dielectric strength of PVC is _____.
5. The electrical resistance of a super conductor is _____.
6. The example for soft super conductor is _____.
7. The longest nano tubes can have length of the order of _____.
8. A superconducting material behaves like a perfect _____ material.

(8 × 1 = 8)

Part B

Answer any six questions.

Each question carries 2 marks.

9. How does a crystal differ from lattices ?
10. What is fcc structure ?
11. What is Bragg's law ? Explain.
12. Explain spin waves.
13. What is anti-ferromagnetism ?

Turn over

14. State the physical significance of Clausius - Mosotti relation.
15. What is nano structure ?
16. What is Meissner effect ?
17. What is SQUID ?
18. Explain the term "Band Gap Energy".

(6 × 2 = 12)

Part C

*Answer any four questions.
Each question carries 4 marks.*

19. Obtain Miller indices for planes (3a, 3b, 3c).
20. Calculate the wavelength of the x-rays incident at 15° with the plane of the first order reflection from a calcite crystal with inter atomic spacing 3.5\AA .
21. What is paramagnetism ? Obtain Langevin's formula for paramagnetic susceptibility.
22. A superconductivity material has a critical temperature of 3.7K in zero magnetic fields and a critical field of 0.0306 Tesla at 0K. Find the critical field at 2K.
23. What is London Equation in superconductivity ? Explain.
24. Give an account on liquid crystals.

(4 × 4 = 16)

Part D

*Answer any two questions.
Each question carries 12 marks.*

25. Explain briefly the Powder Method for crystal structure analysis.
26. Explain the two dimensional and three dimensional lattice types.
27. Explain the domain theory for ferromagnetism.
28. Explain in detail about Type I and Type II superconductors.

(2 × 12 = 24)